

### Investigation 2a: What Do Particles Look Like?

#### Goals and Objectives

Students will understand that particles in the air can be collected in various ways and observed.

- Students will use the engineering design process to design a method for collecting particle pollution data.
- Students will calculate the rate of particle collection.

#### Time Required

Two 50-minute periods.

You will need a minimum of one class period to design the collectors and a second class period to analyze the results.

#### Standards

##### Grade 6, ACCRS, Mathematics, 6.RP.A.3

Use ratio and rate reasoning to solve real-world and mathematical problems.

##### Grade 7, ACCRS, Mathematics, 7.NS.A.3

Solve real-world and mathematical problems involving the four operations with rational numbers.

##### Grade 6-8, ELA, 6.W.1, 7.W.1, 8.W.1

Write arguments to support claims with clear reasons and relevant evidence.

##### Grades 9-12, ELA, 9-10.W.1, 11-12.W.1

Write arguments to support claims in an analysis of substantive topics or texts, using valid reasoning and relevant and sufficient

#### Materials

- Class set of Investigation 2a: Lab Sheet
- Class set of Investigation 2a: Planning Sheet (Option 1 or Option 2)
- Filter paper, one 12 inch piece for each group
- 3X5 index cards
- Graph paper
- Hole punches, one per group
- Scissors, one per group
- String
- Pipe cleaners
- Craft sticks
- Clear transparent tape (not the invisible kind)
- Glue sticks
- 40X Magnifiers or hand lens, one per group
- Microscopes and glass slides (optional)
- Science journals or notebooks

#### Essential Question

What is in the air we breathe?

## Investigation 2a: Lesson Plan

### Pre-Lab Guiding Questions



Begin the lesson by asking students the following questions:

1. **What gases are in the air we breathe?** Oxygen, nitrogen, water vapor, carbon dioxide methane, ozone, etc.
2. **What else is in the air we breathe?** Water droplets, dust, soot, particles, volatile organic compounds, etc.

### Introduction



1. Divide students into cooperative learning groups. Pass out samples of filter paper to each group. Ask them to discuss what they think the papers might be, what the different colored circles represent and how the circles got onto the paper, etc. Students record their ideas in a science journal or notebook.
2. Tell students: **Particles in the air can have harmful effects on people's health. So, it makes sense that scientists want to better understand what kind of particles are floating around the different spaces in which people spend time.** Collect the filter papers. Students will return to these in Investigation 3b.

### Procedure



Students will work in groups to complete an engineering design project. Their challenge is to design and build an instrument that will collect airborne particles.

**Explain:** One way scientists collect particles in the air is to use something called a **high-volume sampler**. This sampler uses a pump to pull air through a filter that traps particles of a certain size. The longer the pump is run, the greater the number of particles that will get trapped.

Another way to collect particles is to wait for them to settle onto a sticky surface, like a piece of tape. This method works well with coarse particles, since they settle out of the air quickly. Because coarse particles are larger, you can view them using a classroom microscope or hand lens.

There are two options for this investigation.

## Procedure Continued...



**Option #1:** Students follow the step by step directions for building a particle pollution collector using an index card and tape. (See Planning Sheet-Option #1)

**Option #2:** Students design their own particle pollution collectors following the engineering design process outlined on Planning Sheet-Option #2. Students first identify the problem, brainstorm a list of possible solutions, and choose the best one. Then, they draw a diagram of their proposed collector, listing all needed materials. They create a prototype and test it outside. Finally, they evaluate the effectiveness of their collector and revise the design to improve upon it. Option #2 may require additional class time for students to design and refine their collectors.

Regardless of which option is chosen, students will need to give some thought to selecting a location for their collectors, identifying potential sources of particle pollution, as well as any variables that might affect the results of the investigation. It is recommended that students sample the air for a minimum of 24 hours. Students will complete Investigation 2a: Lab Sheet after they have retrieved their particle pollution collectors.

## Data and Observations



- Students will record quantitative data by completing the data table on their lab sheet. Remind them to record the location of the card and the number of small, medium, and large particles they see using either a magnifier or microscope.
- Students will record qualitative data by describing the particles. They may include diagrams or sketches. Students should write a claim statement about the source of the particles and cite specific evidence to support their claim.

## Calculations



Tell students: When a scientist measures particles, he/she uses an air pump to measure the volume of air that passes through a filter. Then, after weighing the particles on a filter, the scientist reports data in  $\mu\text{g/L}$ , or micrograms (mass) of particles per liter of air. Since you can't weigh your particles, and we didn't measure how much air touched your card, we'll have to use another measure: rate of particle collection.

Students follow the steps under the calculations section of their lab sheet to determine the particle collection rate.

## Making Sense of Your Results



Working in cooperative groups, students complete the Making Sense of Your Results section of the lab sheet.

## Post-Lab Discussion



Ask students to report their results to the class during a whole group discussion. Other post-lab discussion questions may include the following:

- What are some indoor particle sources? *Wood smoke, household chemicals, hair, pet hair, carpet fibers, cooking*
- What are some outdoor particle sources? *Vehicles (from both combustion and tire and brake wear), power plants, fires, pollen, construction dust, volcanoes*
- Do cars make particles? *Yes, especially if they have black smoke in their exhaust. Particles also come from the rubber bits that rub off of tires.*
- What kinds of vehicles make the most particles? *Vehicles with unregulated diesel engines, such as old trucks or construction equipment.*

At the conclusion of the lab, students store their particle collectors in a re-sealable plastic bag. They will be needed in Investigation 3a.

## Going Further



A scientist used a filter weighing 4.020 grams to collect particles. After pumping 750 mL of air through the filter, it weighed 4.031 grams. What was the concentration of particle mass in the processed air, in milligrams per liter (mg/L)?

$$4.031 \text{ g} - 4.020 \text{ g} = 0.011 \text{ g or } 11 \text{ mg}$$

$$750 \text{ mL} = 0.75 \text{ L volume}$$

$$11 \text{ mg} / 0.75 \text{ L} = 14.7 \text{ mg/L}$$

## Going Further Continued...



The concentration of particles in the air is really small, so the mass units need to be really small as well. We commonly use micrograms instead of milligrams when measuring particle pollution concentrations. We also need a lot of air, so we use a cubic meter instead of a liter of air. Imagine a box of air that is one meter on each side. If my filter paper weighs 2.011 milligrams when clean and 2.023 mg after I pump one cubic meter of air through it, what would the concentration of particles be in  $\mu\text{g}/\text{m}^3$ ?

*2.023 mg - 2.011 mg = 0.012 mg and there are 1000 micrograms in a milligram, so 12 micrograms of particles per cubic meter of air ( $12 \mu\text{g}/\text{m}^3$ ). This concentration is well below the current 24-hour standard of  $35 \mu\text{g}/\text{m}^3$ .*

## On a Personal Note



After seeing all the coarse particles collected in this lab, you decide to wear a dust mask from now on. Will that protect you from the most dangerous particles? Why or why not?

*No, the smallest particles are the most dangerous and a dust mask might not stop them. Students might say "Yes, if you buy a dust mask good enough to stop the fine and ultrafine particles that pose a health risk."*

## Enhancements



Collect particles from specific sources, such as near a construction site, a busy road, a house with smoke coming out of the chimney or a farm. Examine the samples under a magnifier or microscope.

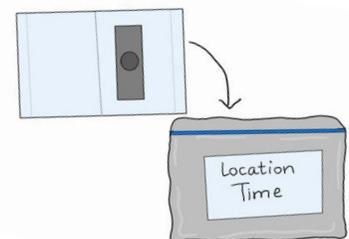
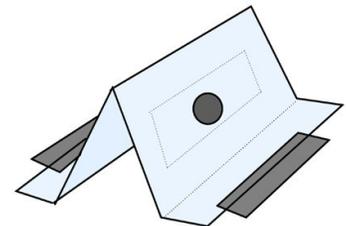
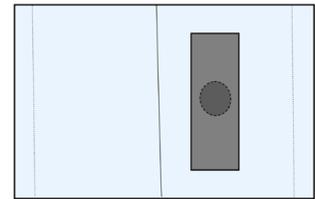
Name: \_\_\_\_\_

## Investigation 2a: Planning Sheet-Option 1

### Procedure



1. Take two 3x5 index cards and fold each one in half.
2. Punch a hole in the middle of one side of each card. Be sure the holes are no bigger than the width of your tape, so that the tape will cover the entire hole.
3. Fold over 1/2 inch on each end of the index cards to make the cards stand.
4. Place your tape over the hole on each card, with the sticky side of the tape facing outside the fold. Clear tape works best.
5. Put each of your cards in different locations for a measured amount of time. Secure the card to the location with tape. Record on your cards where you put them and how many minutes they sat there. Think carefully about where to put your cards and what type of particles you expect to collect.
6. Take note of the surroundings that might produce particles such as wood stoves, barbeques, cars, lawn mowers, air conditioning units, street sweepers, trains, buses, trees with pollen, animals, dusty fields, pet fur and evaporating liquids such as gasoline. Draw a sketch of what is near each of your cards.
7. If you are collecting particles away from the classroom, put each of your cards in its own plastic bag to keep them protected between your sampling time and your class period.



### Sketch of Location

Name: \_\_\_\_\_

# Investigation 2a: Planning Sheet-Option 2

## Procedure



**1. Identify the Problem**

**2. Brainstorm Your Ideas**

**4. Test Your Collector! Sketch the Location**

**3. Choose the Best Solution - Draw a Diagram**

**List all Needed Materials**

**5. Improve - How can you refine your design?**

## Investigation 2a: Lab Sheet

### Data



Examine a clean piece of tape first. This is your control. Now examine your exposed tape. Count the number of particles you see by size and record the numbers in the table below.

If you are using a microscope, gently peel the exposed tape off of the card and stick it to a

Location	Small Particles	Medium Particles	Large Particles

### Observations



1. Describe some of your largest particles here, include drawings.

2. Can you tell where they came from? How do you know? Cite specific evidence to support your claim.

### Investigation 2a: Lab Sheet

#### Calculations



When scientists measure particles, they use an air pump to measure the volume of air. Then, after weighing the particles on a filter, the scientist reports data in  $\mu\text{g/L}$ , or micrograms (mass) of particles per liter of air. Since you can't weigh your particles, and we didn't measure the air that touched your card, we'll have to use another measure: rate of particle collection.

1. What is the total number of particles that you collected? \_\_\_\_\_
2. What is the total number of minutes that your card was exposed? \_\_\_\_\_  
(If you collected for several hours, multiply the number of hours by 60 to find the number of minutes.)
3. At what rate (in particles per minute) did you collect particles?  
\_\_\_\_\_ particles  $\div$  \_\_\_\_\_ minutes = \_\_\_\_\_ particles/minute  
(Divide the number of particles by the number of minutes your collector was exposed to the air.)

#### Making Sense of Your Results



1. Can you think of other methods for testing the air for coarse particles?
  
  
  
  
  
  
  
  
  
  
2. Were there certain places where you or your classmates collected more particles than you collected in other places? Where? What was the source of the particulate pollution?